## Enantioselective Total Synthesis of (–)-Dehydrobatzelladine C

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## ABSTRACT

 $NH_2$  $\ddot{N}H_2$ 

(-)-dehydrobatzelladine C

The oxidation of two tethered Biginelli adducts was examined as a potential key step in total syntheses of highly oxidized batzelladine and crambescidin alkaloids. Although angular hydroxyl substitution could not be introduced, dehydrogenation was readily accomplished. This latter conversion is a key step in the first total synthesis of dehydrobatzelladine C.

A variety of structurally intricate guanidine alkaloids are present in marine sources.<sup>1</sup> Among the most notable of these are the crambescidin<sup>2</sup> and batzelladine<sup>3</sup> alkaloids, which have been isolated primarily from sponges belonging to the orders Poecilosclerida and Axinellida.<sup>1</sup> Diverse biological activities have been reported for these secondary metabolites, including cytotoxicity toward several cancer cell lines, antifungal and antiviral activities, and inhibition of HIV-1 fusion.<sup>1,4</sup> The novel structures of these marine alkaloids have inspired the development of many strategies for assembling polycyclic guanidines that contain the octahydro-5,6,6a-triazaacenaphthalene (1) and hexahydro-5,6,6a-triazaacenaphthalene (2)

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moieties common to the crambescidin and batzelladine alkaloids.<sup>1,5,6</sup>

Intramolecular variants of the Biginelli condensation<sup>7</sup> have been developed in our laboratories for the synthesis of crambescidin<sup>6</sup> and batzelladine<sup>8</sup> alkaloids as well as simplified congeners.<sup>9</sup> In this communication, we report our preliminary efforts to access rare members of the crambescidin family that possess either angular hydroxyl substitution on the central tricyclic octahydro-5,6,6a-triazaacenaphthalene fragment such as crambescidin 816 (**4**) or a tetrahy-

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Francisco, CA, 94080.

<sup>(1)</sup> For reviews, see: Berlinck, R. G. S. Nat. Prod. Rep. 2002, 19, 617 and earlier reviews in this series.

<sup>(2)</sup> For the initial reports, see: (a) Kashman, Y.; Hirsh, S.; McConnell, O. J.; Ohtani, I.; Kusumi, T.; Kakisawa, H. J. Am. Chem. Soc. 1989, 111,

<sup>(3) (</sup>a) Patil, A. D.; Kumar, N. V.; Kokke, W. C.; Bean, M. F.; Freyer, A. J.; De Brosse, C.; Mai, S.; Truneh, A.; Faulkner, D. J.; Carte, B.; Breen, A. L.; Hertzberg, R. P.; Johnson, R. K.; Westley, J. W.; Potts, B. C. M. J. Org. Chem. **1995**, 60, 1182. (b) Patil, A. D.; Freyer, A. J.; Taylor, P. B.; Carte, B.; Zuber, G.; Johnson, R. K.; Faulkner, D. J. J. Org. Chem. **1997**, 62. 1814.

<sup>(4)</sup> Chang, L.; Whittaker, N. F.; Bewley, C. A. J. Nat. Prod. 2003, 66, 1490.

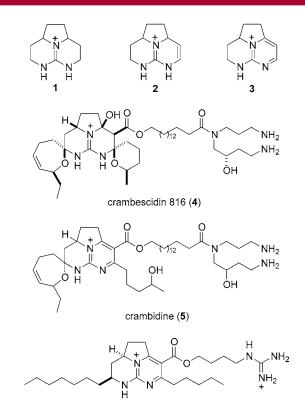
<sup>(5)</sup> For total syntheses appearing since the most recent reviews, see; (a) Nagasawa, K.; Georgeiva, A.; Koshino, H.; Nakata, T.; Kita, T.; Hashimoto, Y. *Org. Lett.* **2002**, *4*, 177. (b) Ishiwata, T.; Hino, T.; Koshino, H.; Hashimoto, Y.; Nakata, T.; Nagasawa, K. *Org. Lett.* **2002**, *4*, 2921. (c) Nagasawa, K.; Ishiwata, Y.; Hasimoto, Y.; Nakata, T. *Tetrahedron Lett.* **2002**, *43*, 6383. (d) Moore, C. G.; Murphy, P. J.; Williams, H. L.; McGown, A. T.; Smith, N. K. *Tetrahedron Lett.* **2003**, *44*, 251.

<sup>(6) (</sup>a) Overman, L. E.; Rabinowitz, M. H.; Renhowe, P. A. J. Am. Chem. Soc. 1995, 117, 2657. (b) Overman, L. E.; Rabinowitz, M. H. J. Org. Chem. 1993, 58, 3235. (c) Coffey, D. S.; McDonald, A. I.; Overman, L. E.; Rabinowitz, M. H.; Renhowe, P. A. J. Am. Chem. Soc. 2000, 122, 4893. (d) Coffey, D. S.; Overman, L. E.; Stappenbeck, F. J. Am. Chem. Soc. 2000, 122, 4904. (e) Coffey, D. S.; McDonald, A. I.; Overman, L. E.; Stappenbeck, F. J. Am. Chem. Soc. 1999, 121, 6944.

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<sup>(8) (</sup>a) Franklin, A. S.; Ly, S. K.; Mackin, G. H.; Overman, L. E.; Shaka, A. J. J. Org. Chem. **1999**, 64, 1512. (b) Cohen, F.; Overman, L. E.; Sakata, S. K. L. Org. Lett. **1999**, 1, 2169. (c) Cohen, F.; Overman, L. E. J. Am. Chem. Soc. **2001**, 123, 10782.

dro-5,6,6a-triazaacenaphthalene moiety (**3**) such as that found in crambidine (**5**)<sup>10</sup> and dehydrobatzelladine C (**6**).<sup>11</sup> The total synthesis of (–)-dehydrobatzelladine C (**6**) achieved during these investigations constitutes the first total synthesis of a member of the crambescidin or batzelladine families that contains the tetrahydro-5,6,6a-triazaacenaphthalene fragment.

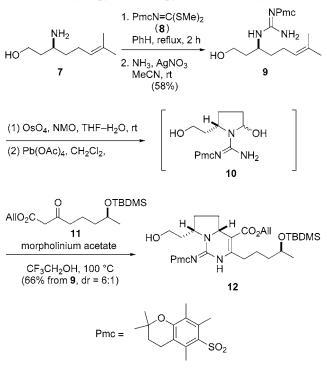


dehydrobatzelladine C (6)

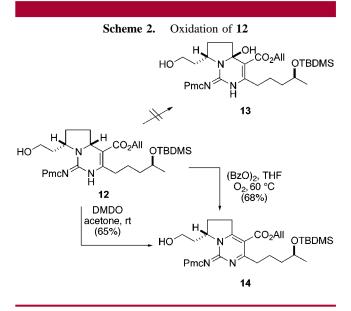
**Figure 1.** Representative polycyclic guanidine alkaloids containing hydro-5,6,6a-triazaacenaphthalene fragments.

1-Oxo- or 1-iminohexahydropyrrolo[1,2-*c*]pyrimidine carboxylic esters were key intermediates in our earlier total syntheses of crambescidin alkaloids.<sup>6,7</sup> In this investigation, we first prepared an *N*-sulfonyl-protected version of the latter intermediate, **12**,<sup>12</sup> which we hoped could be selectively hydroxylated at the angular carbon  $\beta$  to the ester substituent (Scheme 1). The 2,2,5,7,8-pentamethylchroman-6-sulfonyl (Pmc) guanidine fragment was introduced in two steps by sequential reaction of amino alcohol **7**<sup>6c</sup> with carboimidodithioate **8** and ammonia.<sup>13</sup> Oxidative cleavage of the double bond of **9**, followed by tethered Biginelli condensation of the resulting crude product **10** with  $\beta$ -ketoester **11**, provided sulfonyliminopyrrolopyrimidine **12** in 66% yield from **9**.<sup>14</sup>

We explored oxidation of **12** with a variety of agents. Unfortunately, all attempts to convert **12** to **13** by reaction **Scheme 1.** Synthesis of Pmc-Protected 1-Imino-hexahydropyrrolo[1,2-*c*]pyrimidine Carboxylic Ester **12** 



of the former with reagents such as  $Pb(OAc)_4$ ,  $H_2O_2$ , t-BuO<sub>2</sub>H, SeO<sub>2</sub>, RuO<sub>2</sub>, or DDQ, under a variety of reaction conditions, were unsuccessful (Scheme 2).<sup>15</sup> Only the reaction



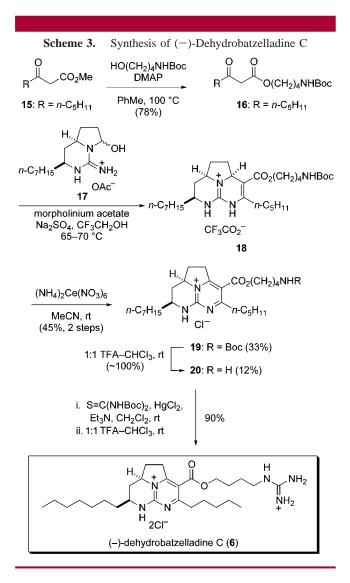
of **12** with dimethyldioxirane or benzoyl peroxide delivered a clean product, in these cases the didehydro derivative **14**. This dehydrogenation of **12** suggested a straightforward way to synthesize alkaloids such as **5** or **6** that contain a 2-aminopyrimidinium unit.<sup>16</sup>

Dehydrobatzelladine C ( $\mathbf{6}$ ) was chosen as our initial target (Scheme 3). The total synthesis of  $\mathbf{6}$  began by DMAP-

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W. M. J. Nat. Prod. 2000, 63, 193.
(12) McDonald, A. I.; Overman, L. E. J. Org. Chem. 1999, 64, 1520.

<sup>(12)</sup> McDonau, A. I., Overman, E. E. J. Org. Chem. 1999, 04, 1520 (13) Heizmann, G.; Felder, E. R. *Peptide Res.* **1994**, 7, 328.



catalyzed transesterification<sup>17</sup> of methyl 3-oxooctanoate (**15**) with *N*-Boc-protected 4-aminobutanol to provide  $\beta$ -keto ester **16** in 78% yield. Biginelli condensation of 2.7 equiv of **16** with enantiopure guanidine aminal **17**, which is available in 10 steps from commercial materials,<sup>18</sup> took place in CF<sub>3</sub>-CH<sub>2</sub>OH at 65–70 °C with high stereoselectivity (>10:1) to

form hexahydro-5,6,6a-triazaacenaphthalene 18. As this product was difficult to separate from residual  $\beta$ -ketoester,<sup>19</sup> a mixture of these compounds was oxidized directly with 1 equiv of cerric ammonium nitrate at room temperature in acetonitrile.<sup>20</sup> Purification of this product by reverse-phase preparative HPLC then provided 19 as the hydrochloride salt in 33% overall yield and 12% of the corresponding primary amine 20. This latter product is readily generated from Boc derivative 19 by reaction with a 1:1 mixture of TFA/CHCl<sub>3</sub>. Following extensive experimentation, it was found that primary amine 20 was best elaborated to the corresponding guanidine by first condensing the crude amine with N,N'di(tert-butoxycarbonyl)thiourea followed by removal of the Boc groups with acid. Purification of this product by reversephase preparative HPLC provided pure (-)-dehydrobatzelladine C (6) as its dihydrochloride salt in 90% yield. The synthetic product,  $[\alpha]_D$  –88 (c 0.23, MeOH), showed <sup>1</sup>H NMR spectra consistent with that of the natural product.<sup>21</sup> Moreover, <sup>13</sup>C NMR spectra of the diacetate salt of 6 matched perfectly ( $\pm 0.1$  ppm for all signals) with those reported for the natural alkaloid.<sup>11</sup> The optical rotation of the marine isolate was not reported,<sup>11</sup> precluding comparison of this property.

In summary, a wide variety of 1-iminohexahydropyrrolo-[1,2-c]pyrimidine carboxylic esters can be prepared by tethered Biginelli condensations.<sup>6,7</sup> These products can be selectively dehydrogenated to generate congeners containing a 2-aminopyrimidinium moiety. This oxidation is a key step in the first total synthesis of dehydrobatzelladine C (**6**).

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**Supporting Information Available:** Experimental procedures and characterization data for 2,2,5,7,8-pentamethylchroman-6-sulfonamide, *S*,*S*-dimethyl *N*-(2,2,5,7,8-pentamethyl-chroman-6-sulfonyl)carbonimidodithioate (**8**), and new compounds reported in Schemes 1–3, as well as <sup>1</sup>H and <sup>13</sup>C NMR spectra of synthetic (–)-dehydrobatzelladine C (**6**). This material is available free of charge via the Internet at http://pubs.acs.org.

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<sup>(14)</sup> The trans epimer of **12** was isolated in 8% yield. The relative configuration of these epimers was assigned by analogy to closely related Biginelli products whose configuration had been established rigorously by chemical correlation.<sup>12</sup>

<sup>(15)</sup> These reactions either returned **12** or resulted in the formation of intractable mixtures.

<sup>(16)</sup> There are several examples of the conversion of simple Biginelli adducts to pyrimidines; see: (a) Kappe, O. C. *Acc. Chem. Res.* **2000**, *33*, 879. (b) Kappe, O. C. *Org. React.* **2004**, *63*, in press.

<sup>(17)</sup> Taber, D. F.; Amedio, J. C.; Patel, Y. K. J. Org. Chem. 1985, 50, 3618.

<sup>(18)</sup> For the preparation of the enantiomer of 17, see ref 8a.

<sup>(19)</sup> Tethered Biginelli adduct **18** could be isolated in pure form (48% yield) by preparative reverse-phase HPLC.

<sup>(20)</sup> This oxidation can also be accomplished in similar yield with benzoyl peroxide; however, purification of the product is more cumbersome.

<sup>(21)</sup> We thank Professor Braekman for a copy of the 600 MHz <sup>1</sup>H NMR spectrum of authentic dehydrobatzelladine C.